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Lecture 34 . Kinetics I . Worksheet . 1) The data below shows the change in concentration of dinitrogen pentoxide over time at 330 K, according to the following process.  $2 N_2O_5(g) \rightarrow 4 NO_2(g) + O_2(g)$  [N<sub>2</sub>O<sub>5</sub>] Time (s) 0.100 0.00 0.066 200.00 0.044 400.00 . a. Find the rate of disappearance of N<sub>2</sub>O<sub>5</sub> from t = 0s to t = 200s. b.

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