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3 react according to the above equation, how many moles of NO will be consumed? 8. Propyne gas can be used as a fuel. The combustion reaction of propyne can be represented by the following equation:  $C_3H_4(g) + 4O_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  a. Write all the possible mole ratios in this system.

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4 mol O 2:1 mol C 3H 4; 3 mol CO 2:1 mol C 3H 4; 2 ...

### **mc06se cFMs r i-vi**

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### **Modern Chemistry Chapter 5 Section 3 Review Answers**

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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SECTION 3 continued 5. Explain the following properties of solids by describing what is occurring at the atomic level. a. Metallic solids conduct electricity well, but covalent network solids do not. Metals have many electrons that are not bound to any one atom; therefore they are able to move throughout the crystal.

### **10 States of Matter - Ms. Agostine's Chemistry Page**

Modern Chemistry 84 . Name Section Quiz, continued Class Date 5. In some instances, the concentration of a solution is expressed as molality instead of molarity because a. molality is easier to calculate. b. molarity applies only to solid-liquid solutions.

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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