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## **HW Solutions Section 1**

1.11 Both Vitamin B12 and Visudyne are porphyrin-based. 1.13 A natural product is a compound produced by a living organism. 1.15 (a) The Haber process combines hydrogen and nitrogen to make ammonia. Ammonia is used to make fertilizer. (b) In the Bohr model, a hydrogen atom consists of an electron in a circular orbit about a proton.

## **Answers to Selected Textbook Questions - Nelson**

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Solution:  $W_{net} = F_{net} \Delta d$   $F_{net} = W_{net} / \Delta d = 127\,400 \text{ J} / 82 \text{ m} = 1553.6 \text{ J/m} = 1554 \text{ N}$   $F_{net} = 1.6 \times 10^3 \text{ N}$   
Statement: The net force acting on the car is  $1.6 \times 10^3 \text{ N}$ . 2. Given:  $m = 52 \text{ kg}$ ;  $v_i = 11 \text{ m/s}$ ;  $v_f = 0 \text{ m/s}$ ;  $\Delta d = 8.0 \text{ m}$  Required:  $F_{net}$  Analysis:  $W_{net} = \frac{1}{2} m v_f^2 - \frac{1}{2} m v_i^2$ ;  $W_{net} = F_{net} \Delta d$  (a)  $h$

en tf o rc ska : Solution:  $W_{\text{net}} = mv f^2$  ...

### **Section 5.2: Energy Tutorial 2 Practice, page 232 Tutorial ...**

11(s), (sucrose) because it dissolves well, to produce a nonconducting, neutral solution. Chemical 2 is the ionic compound, KCl (s), because it dissolves well, to produce a conducting, neutral solution. Chemical 4 is the ionic hydroxide, Ba(OH)<sub>2</sub>(s), because it dissolves well, to produce a conducting, basic solution.

### **8.1 UNDERSTANDING ACIDS AND BASES**

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11. (a) The high/low solubility cutoff point is 0.10 mol/L at SATP. (b) The 0.10 mol/L cutoff is useful because most laboratory solutions are within an order of magnitude of this

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